



Molecular structure monitoring  
by vibrational spectroscopy

I. Aplicații ale spectroscopiei vibraționale în studiul moleculelor  
de interes farmaceutic

tiamina, papaverina, metoclopramida

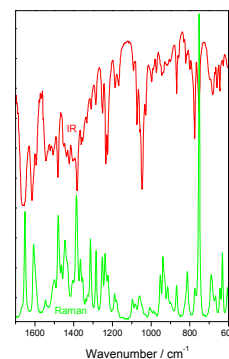
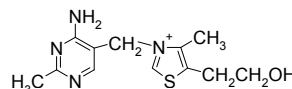
II. Studiu Raman și SERS asupra complexelor metalice

• Scopul studiului

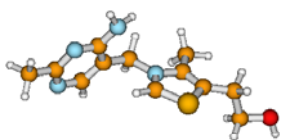
- caracterizarea structurii moleculare prin modurile vibraționale
- atribuirea modurilor vibraționale prin calcule teoretice
- monitorizarea speciilor moleculare la diferite pH-uri
- monitorizarea speciilor moleculare adsorbite la diferite pH-uri
- geometria de adsorbție pe suprafețele metalice

Tiamina – vitamina B<sub>1</sub>

Spectrele Raman și IR  
ale tiaminei în stare solidă

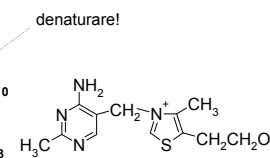
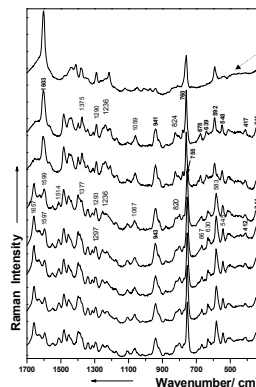


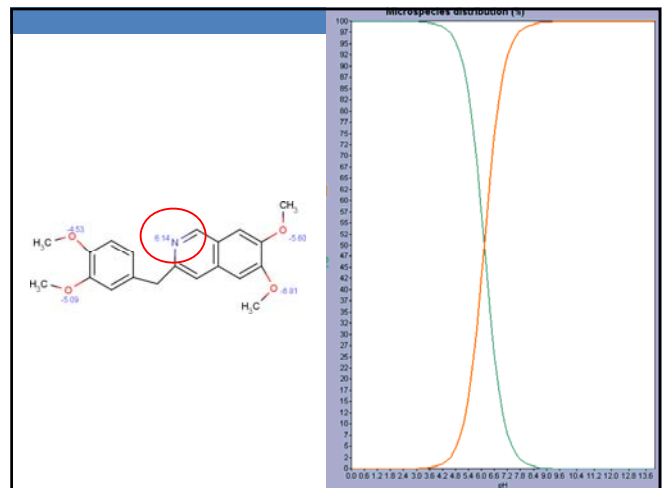
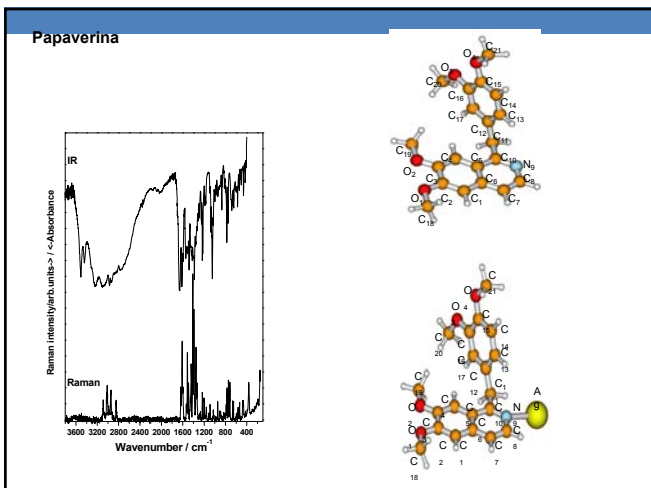
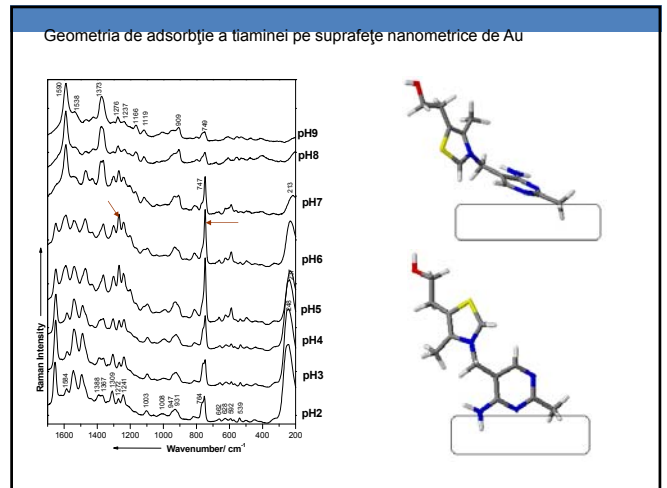
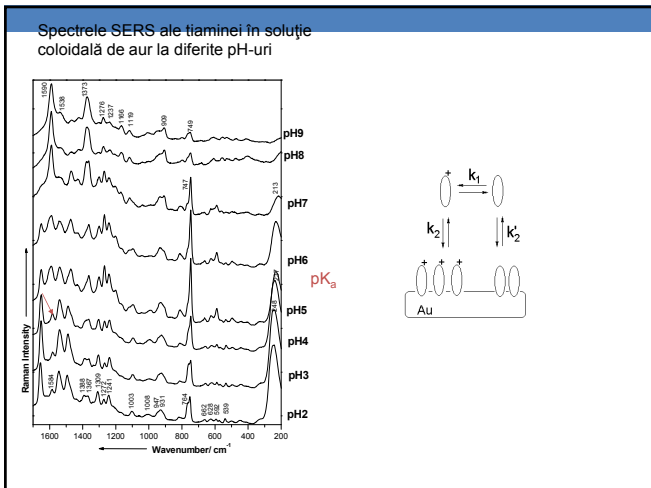
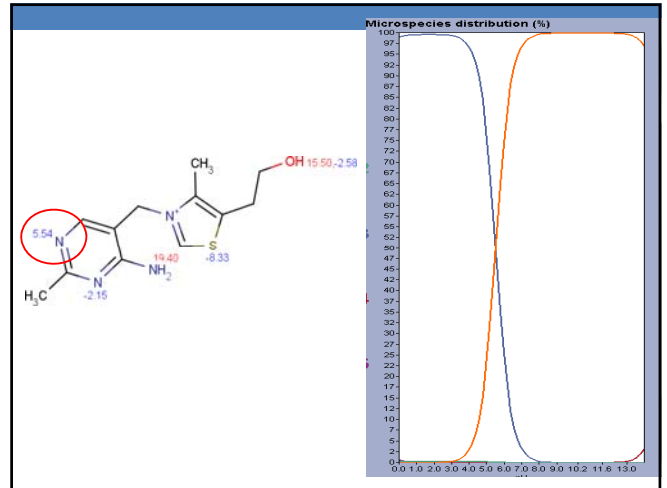
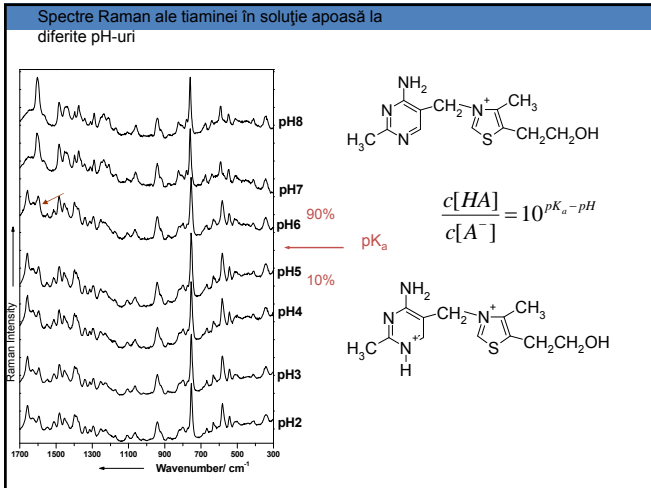
geometria optimizată



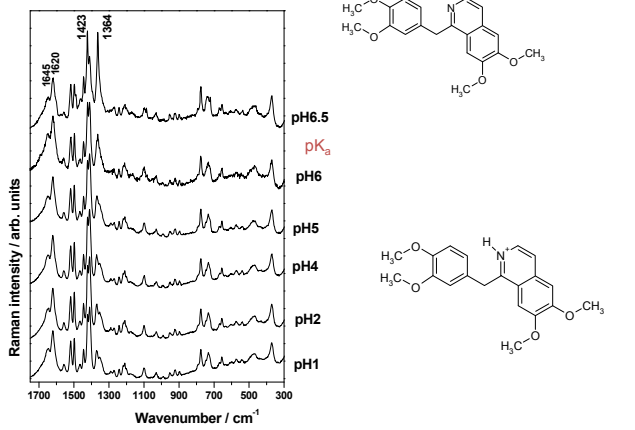
FT-Raman	FTIR	Calc 1	Calc 2	Assignment
3276s		327	327	$\nu_1(\text{ring1})$
3044		332	332	$\nu_2(\text{ring2})$
4030w	4110w	426	426	$\nu_3(\text{ring1}), \nu_4(\text{ring2})$
5180w	5180w	527	527	$\nu_5(\text{ring1})$
548s	5480m	553	552	$\nu_6(\text{NH})$
608w	5950w	605	604	$\nu_7(\text{ring2}), \nu_8(\text{CH2ring1})$ bending
632m	6090w	632	631	$\nu_9(\text{ring1}), \nu_{10}(\text{ring2})$
644w	6440m	647	643	$\nu_{11}(\text{ring1})$
677w	6700w	677	673	$\nu_{12}(\text{ring1})$
689m	6810m	699	699	$\nu_{13}(\text{ring1}-\text{C}-\text{C})$
729	7200w	735	732	$\nu_{14}(\text{ring1})$ bending
814s	8000m	800	800	$\nu_{15}(\text{ring1}), \nu_{16}(\text{ring2})$
869w	8600m	872	872	$\nu_{17}(\text{C})$
903s	8990s	907	902	$\delta(\text{CH}_2-\text{CH}_2)$ bending
915w	9100w	915	912	$\nu_{18}(\text{ring1}-\text{CH}_2)$
930m	9340s	940	932	$\nu_{19}(\text{ring2})$ bending
1043s	10470	1056	1047	$\nu_{20}(\text{C})$
1082w	10720m	1092	1089	$\delta(\text{NH})$
1096w	10910m	1094	1090	$\delta(\text{CH}_2-\text{CH}_2-\text{CH}_2)$ ring1
1177w	11800m	1146	1141	$\beta(\text{ring2})-\nu(\text{CHC})-\nu(\text{CH}_2)$
1180w	11850m	1194	1192	$\nu_{21}(\text{C})$
1220m	12250	1219	1218	$\delta(\text{C}-\text{H})$
1230m	12340	1277	1271	$\nu_{22}(\text{C}-\text{H})$
1251m	12510m	1278	1278	$\delta(\text{CH}_2-\text{CH}_2-\text{CH}_2)$
1284m	12840m	1286	1282	$\nu_{23}(\text{C})$
1353s	13520m	1355	1350	$\delta(\text{CH}_2-\text{CH}_2)$ ring1, ring2
1365m	13610m	1380	1370	$\nu_{24}(\text{ring1})$
1386	13810	1405	1398	$\nu_{25}(\text{ring1})$
1403s	14050m	1406	1406	$\delta(\text{CH}_2-\text{CH}_2-\text{CH}_2)$
1431s	14230m	1435	1427	$\nu_{26}(\text{ring1})$
1445m	14380m	1449	1441	$\delta(\text{CH}_2-\text{CH}_2)$
1479w	14830	1485	1479	$\nu_{27}(\text{ring2})$
1493m	15090m	1499	1493	$\delta(\text{CH}_2-\text{CH}_2)$
1545w	15420m	1542	1538	$\nu_{28}(\text{ring1})$
1606m	16140	1553	1556	$\nu_{29}(\text{ring1})$
1651r	16560	1633	1637	$\nu_{30}(\text{CH}_2-\text{CH}_2)$
2733w	27380w	2794	2782	$\nu_{31}(\text{C}-\text{H})$ bending
28670s	28530s	2997	2984	$\nu_{32}(\text{C}-\text{H})$
2920s	29490s	3049	3003	$\nu_{33}(\text{CH}_2-\text{CH}_2)$
2966w	29690w	3074	3037	$\nu_{34}(\text{C}-\text{H})$ bending
30050s	31010s	3119	3103	$\nu_{35}(\text{CH}_2)$ ring1, ring2
	34420s	3496	3490	$\nu_{36}(\text{N}-\text{H})$
	35940s	3594	3548	$\nu_{37}(\text{C}-\text{H})$

Spectrele Raman ale tiaminei în soluție apoasă la  
diferite pH-uri

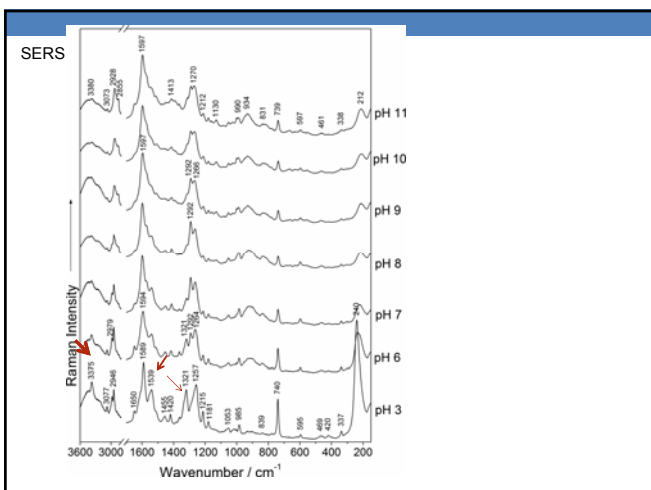
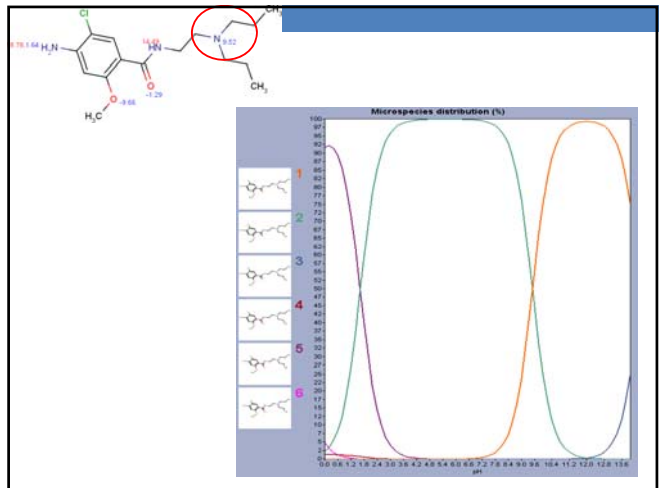
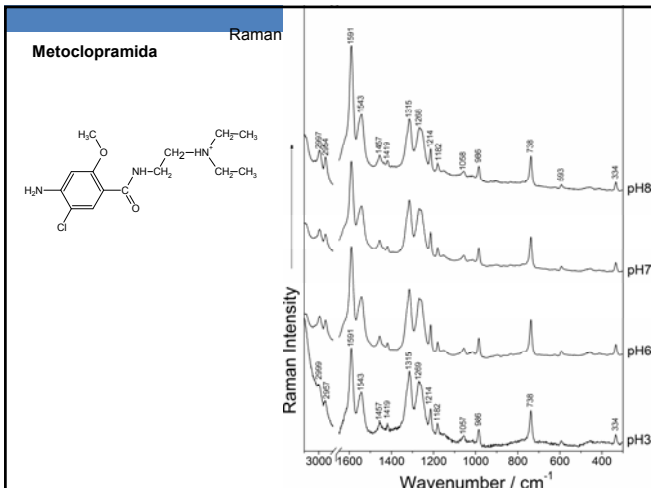
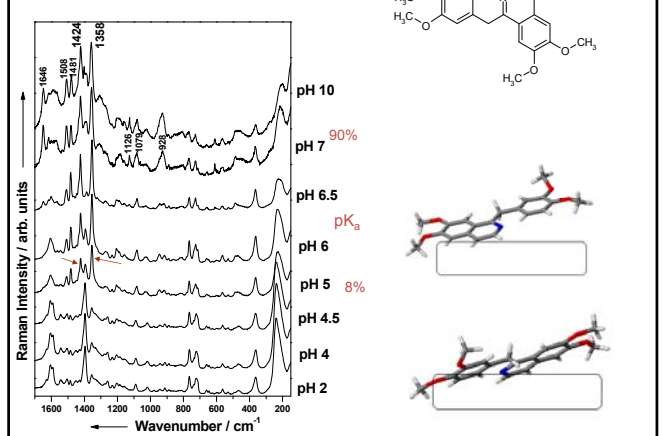




### Spectrele Raman ale papaverinei la diferite pH-uri

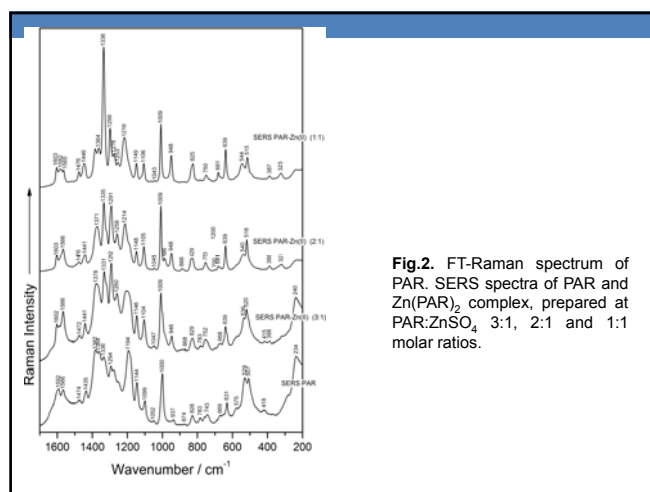
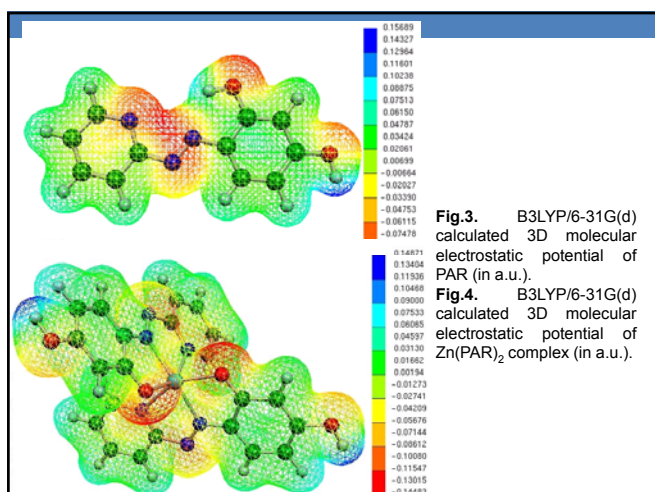
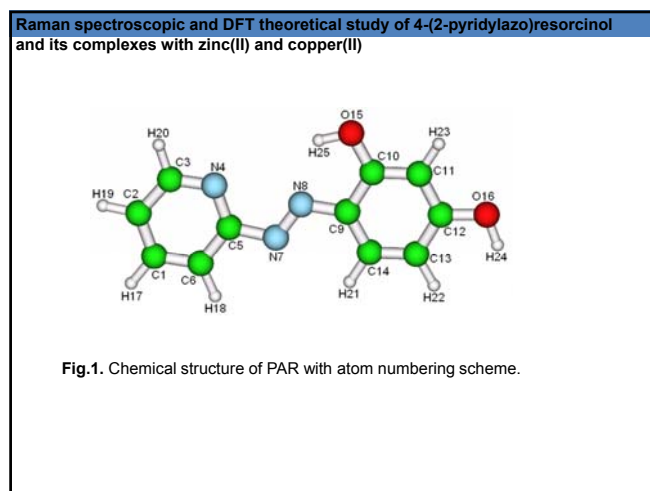
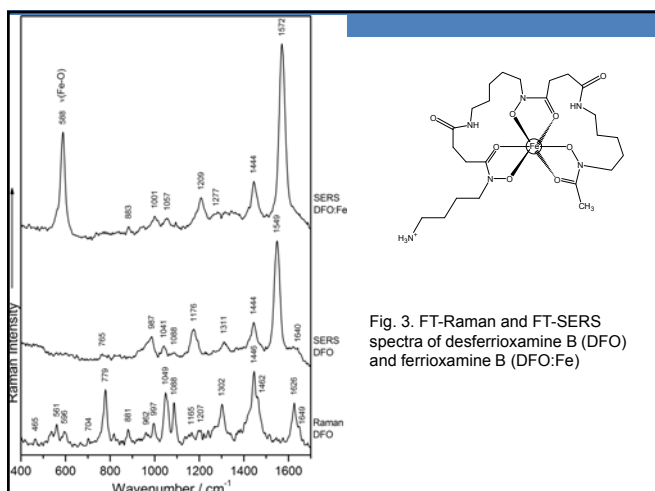
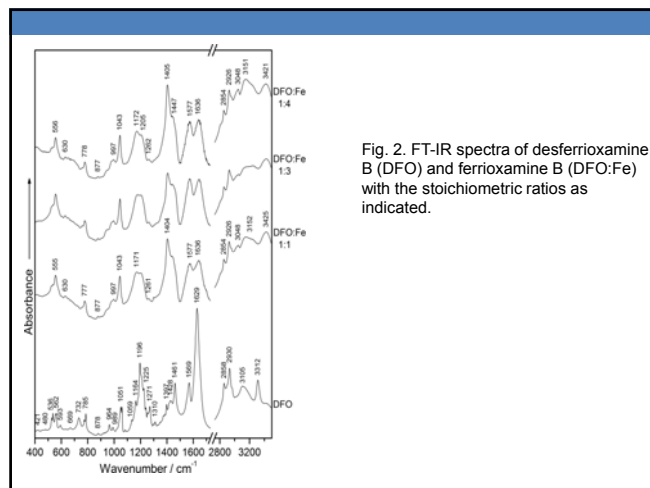
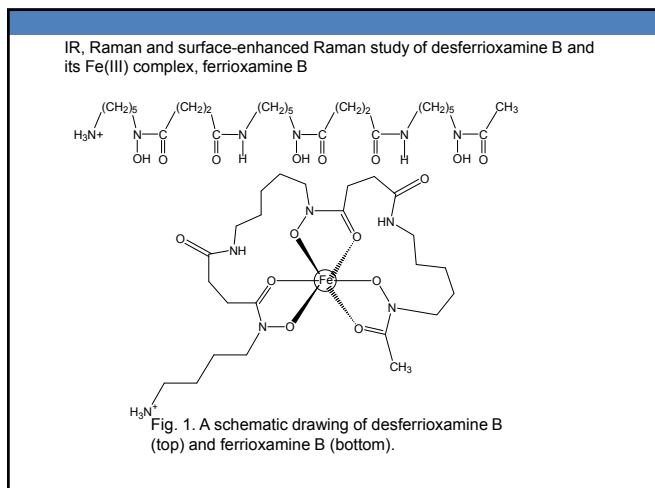


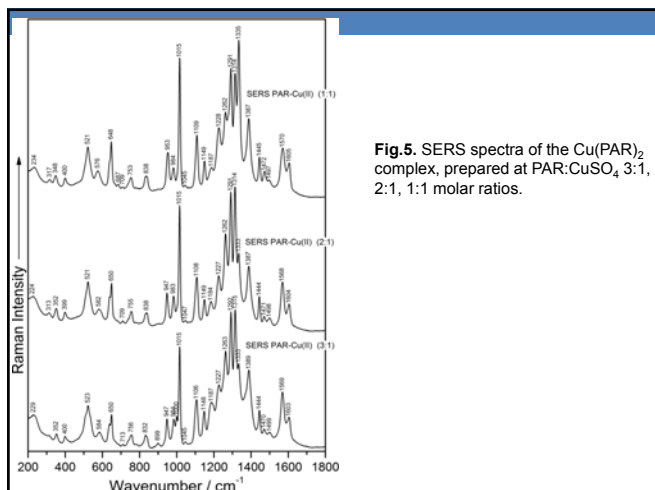
### Spectrele SERS ale papaverinei între pH 2 și pH 10



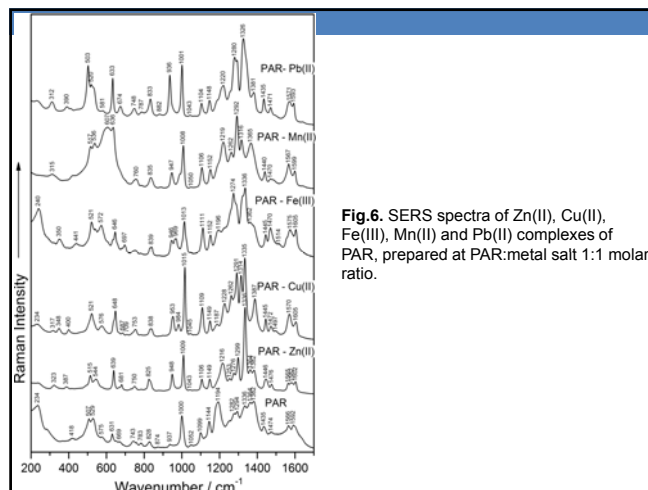
### Concluzii

- ★ spectrele IR, Raman și SERS în diferite condiții experimentale  
→ informații structurale moleculare
- ★ atribuirea modurilor vibraționale prin calcule teoretice
- ★ monitorizarea Raman a speciilor moleculare la diferite pH-uri
- ★ monitorizarea SERS a speciilor moleculare adsorbite la diferite pH-uri
- ★ geometria de adsorbție pe suprafețele metalice

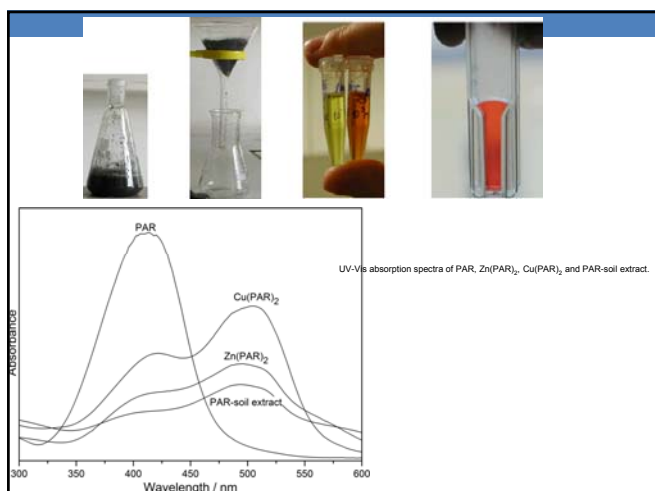




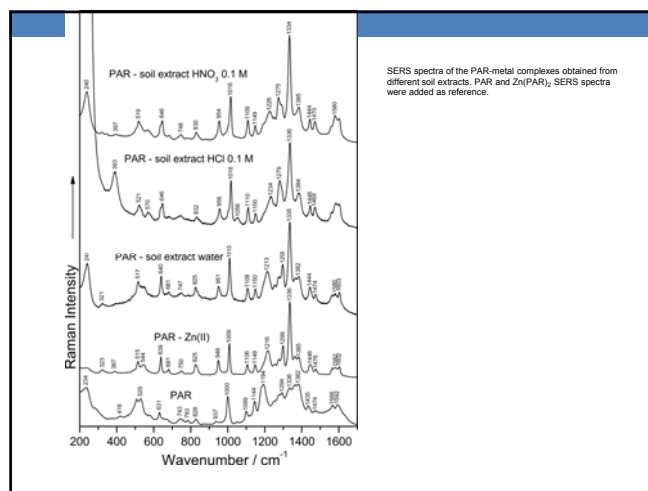
**Fig.5.** SERS spectra of the Cu(PAR)<sub>2</sub> complex, prepared at PAR:CuSO<sub>4</sub> 3:1, 2:1, 1:1 molar ratios.



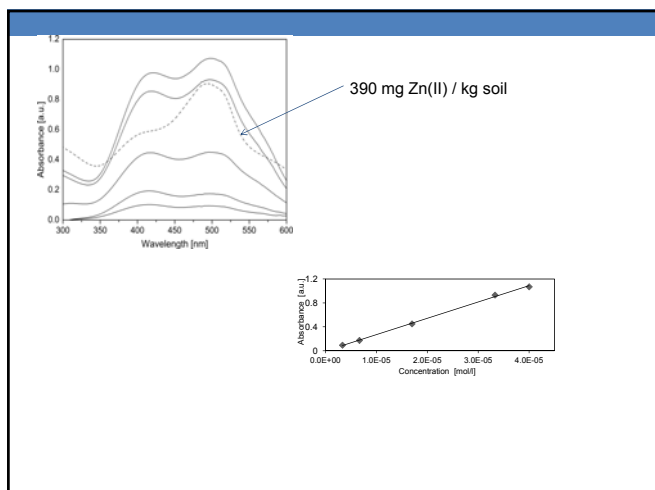
**Fig.6.** SERS spectra of Zn(II), Cu(II), Fe(III), Mn(II) and Pb(II) complexes of PAR, prepared at PAR:metal salt 1:1 molar ratio.



UV-Vis absorption spectra of PAR, Zn(PAR)<sub>2</sub>, Cu(PAR)<sub>2</sub> and PAR-soil extract.



SERS spectra of the PAR-metal complexes obtained from different soil extracts. PAR and Zn(PAR)<sub>2</sub> SERS spectra were added as reference.



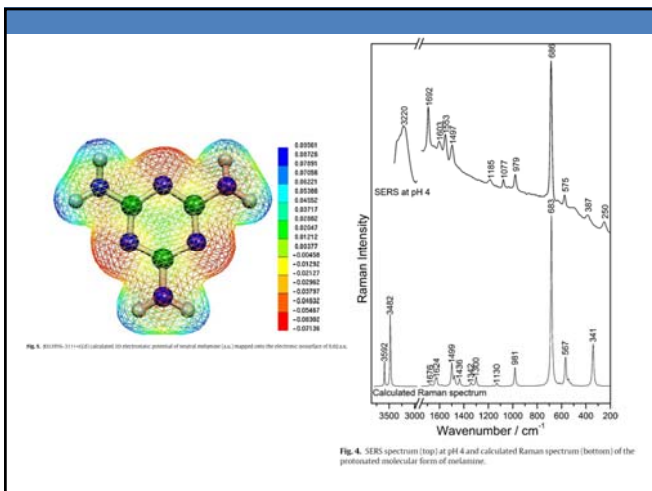
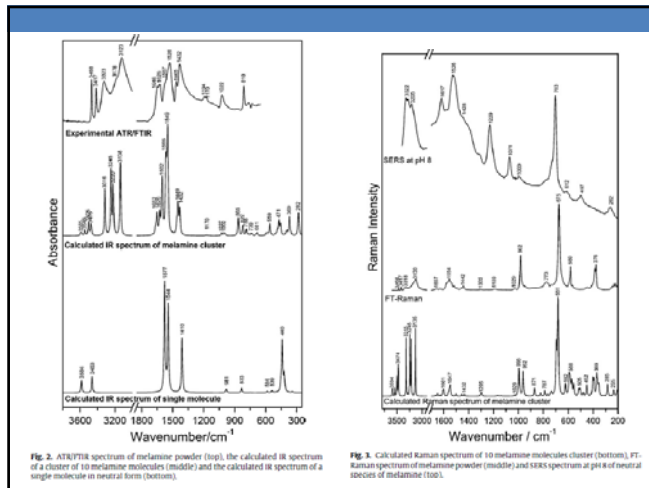
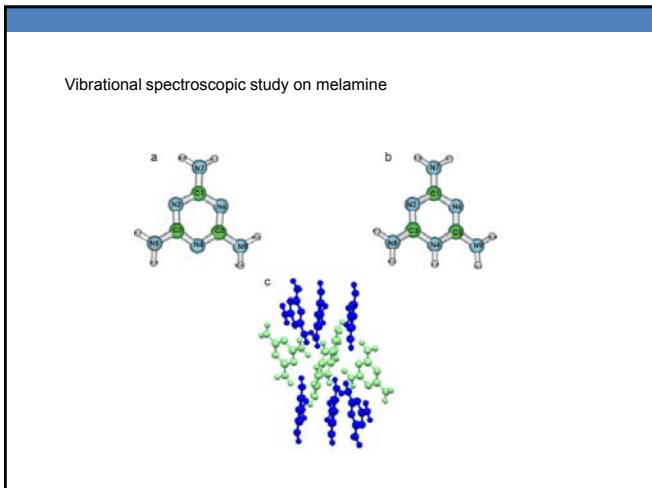
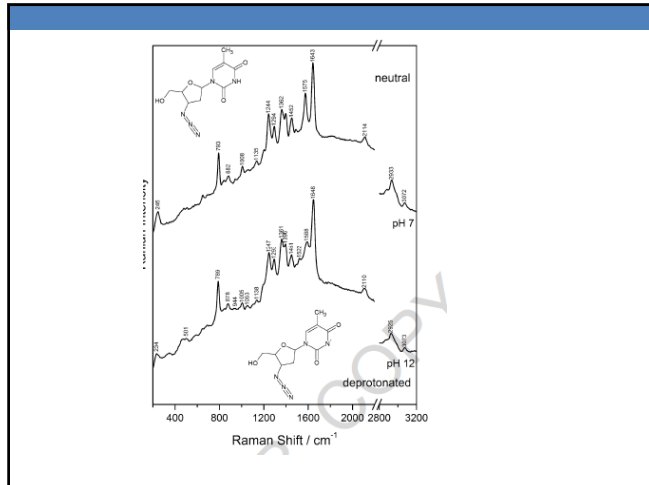
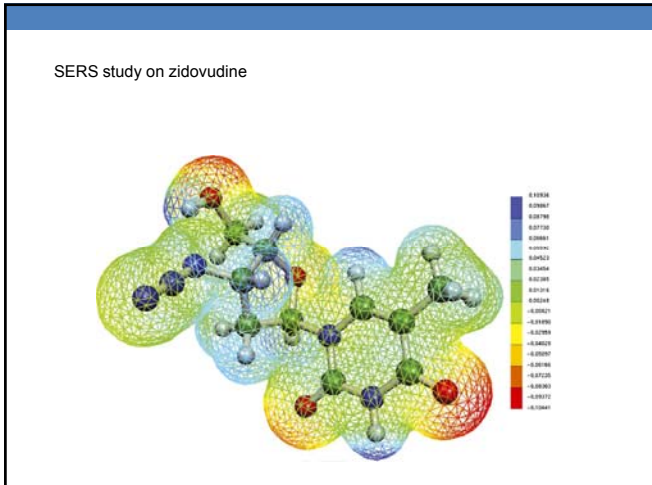
**Conclusions**

The FT-Raman and SERS spectra of PAR, as well as the SERS spectrum of Zn(PAR)<sub>2</sub> complex were safely assigned, due to a good match between experimental and DFT calculated vibrational modes. The SERS band assignment of the Cu(PAR)<sub>2</sub> complex was supposed to be similar to that of Zn(PAR)<sub>2</sub> complex, due to the similitude in band position between the two spectra.

The calculated MEP distributions indicate for the PAR molecule the highest electronegativity localized on the N and O atoms, whereas for the Zn(PAR)<sub>2</sub> complex the negative charge is localized mainly on the O atoms involved in the metal ion coordination, as expected from the deprotonated character of the oxygens.

The SERS spectra of the 3:1 molar ratios show mainly spectral features of the Zn(PAR)<sub>2</sub> or Cu(PAR)<sub>2</sub> complex, but also spectral features of PAR molecules adsorbed to the silver surface, whereas the SERS spectra of the 1:1 molar ratios show exclusively PAR-metal complex spectral features.

As several marker bands are characteristic to each PAR-metal complex, SERS could represent a prospective method for detection of metal ions, like Zn(II), Cu(II), Fe(III), Mn(II) and Pb(II).






## LABEL-FREE DETECTION OF BACTERIA BY USING IN SITU PREPARED SILVER NANOPARTICLES

---

PhD NICOLETA ELENA DINA (MIRCESCU)

### UTI Pathogens Characterization: Challenges

- Is it possible to identify species independent of O-type antigen and strain (for *E. coli*)?
- Can most common uropathogens be identified?

Chem Nucleins Bacterial Cell Wall

Diagram of a Lipopolysaccharide

### In situ 2-step-synthesis of Silver Nanoparticles

Ag<sup>+</sup> AgNO<sub>3</sub> → Ag<sup>+</sup> ⊕

Reductant NH<sub>2</sub>OH·HCl → AgNPs

TEM results

Z. Haibo, D. Yang, N. P. Ivleva, N.E. Mircescu, R. Niessner, C. Haisch, Anal Chem, 86 (3), 2014

### SERS detection of UTI pathogens biomass

- Adsorption of bacterial cells on the coated glass slide by electrostatic forces, no specific receptor required
- 30 min interaction time
- Wash the slide with Millipore water, then SERS spectra acquisition with high reproducibility

**Total time: 2h 30 min**

Conventional cultivation methods → more than 24 h!

### Bacteria discrimination by PCA analysis based on the SERS spectra

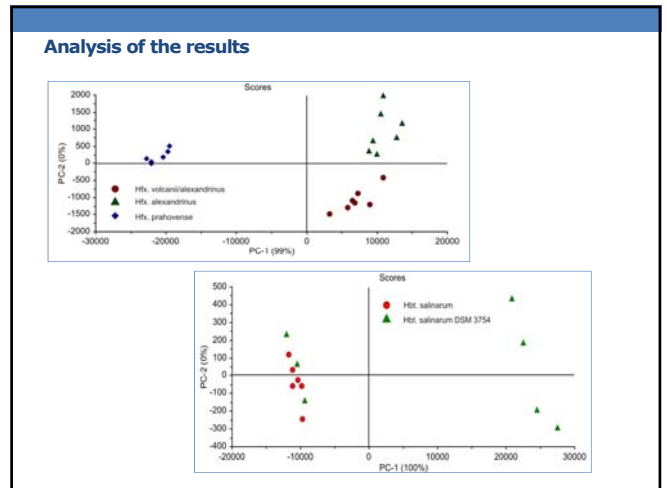
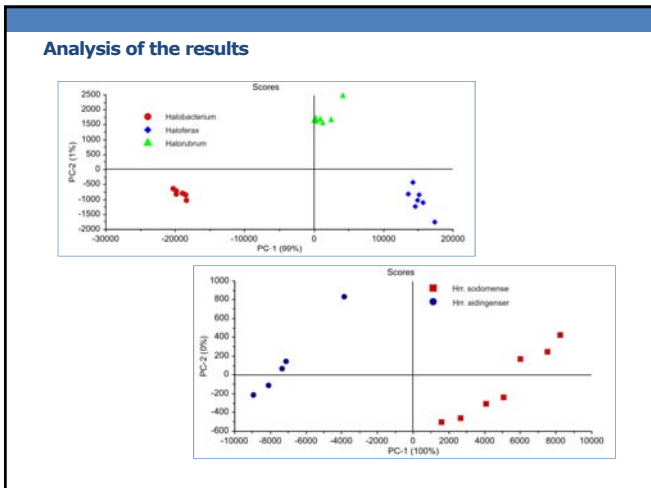
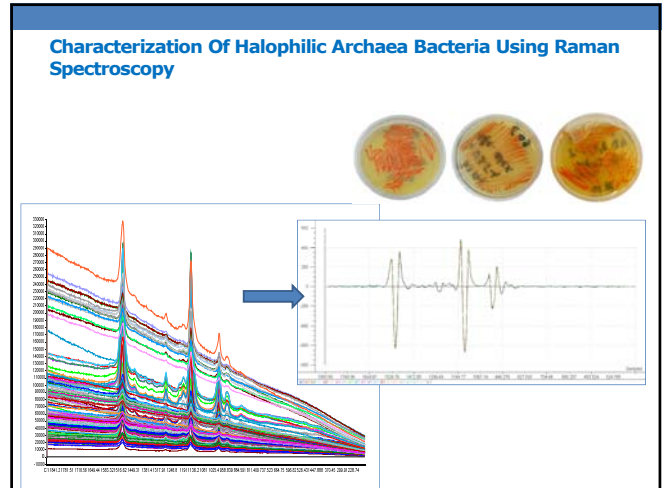
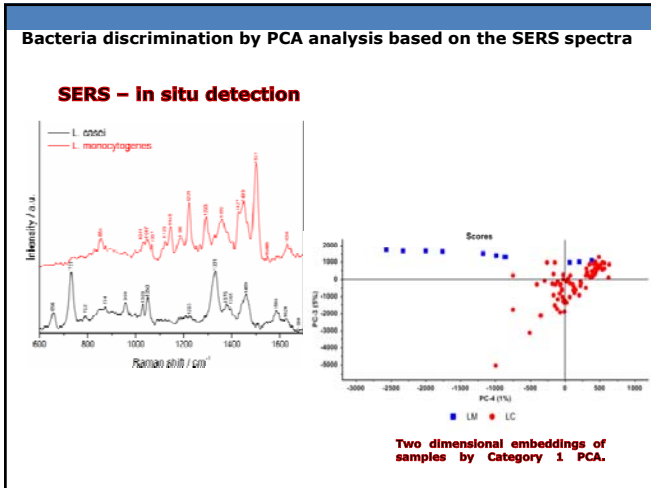
N. E. Mircescu, H. Zhou, N. Leopold, V. Chiş, N. P. Ivleva, R. Niessner, A. Wieser, C. Haisch, Anal Bioanal Chem, 406 (13), 2014;

### Bacteria discrimination by PCA analysis based on the SERS spectra

N. E. Mircescu, H. Zhou, N. Leopold, V. Chiş, N. P. Ivleva, R. Niessner, A. Wieser, C. Haisch, Anal Bioanal Chem, 406 (13), 2014;

### Characterization Of Gram Positive Bacteria Using Raman Spectroscopy

Raman spectra of Gram positive bacteria obtained at 0.05 mW laser power



### Conclusions

- Bacteria discrimination by PCA analysis based on the Raman/SERS spectra is possible independent of the species, O-antigen type, Gram-negative or Gram-positive type.
- A receptor-free SERS based detection method was tested on both Gram-negative and Gram-positive microorganisms for different biomedical purposes.
- The analysis time was significantly reduced by using the in situ SERS active silver nanoparticles.
- The PCA loadings show that the Raman/SERS marker bands are considered as main spectral features and compared in terms of intensity and Raman shifting for discrimination between different bacterial species/strains.

